

# A Mathematical Operations

## A.1 Exponential Notation

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The numbers used in chemistry are often either extremely large or extremely small. Such numbers are conveniently expressed in the form

$$N \times 10^n$$

where  $N$  is a number between 1 and 10, and  $n$  is the exponent. Some examples of this *exponential notation*, which is also called *scientific notation*, follow:

1,200,000 is  $1.2 \times 10^6$  (read "one point two times ten to the sixth power")

0.000604 is  $6.04 \times 10^{-4}$  (read "six point zero four times ten to the negative fourth power")

A positive exponent, as in the first example, tells us how many times a number must be multiplied by 10 to give the long form of the number:

$$1.2 \times 10^6 = 1.2 \times 10 \times 10 \times 10 \times 10 \times 10 \times 10 \times 10 \quad (\text{six tens}) \\ = 1,200,000$$

It is also convenient to think of the positive exponent as the number of places the decimal point must be moved to the *left* to obtain a number greater than 1 and less than 10: If we begin with 3450 and move the decimal point three places to the left, we end up with  $3.45 \times 10^3$ .

In a related fashion, a negative exponent tells us how many times we must divide a number by 10 to give the long form of the number:

$$6.04 \times 10^{-4} = \frac{6.04}{10 \times 10 \times 10 \times 10} = 0.000604$$

It is convenient to think of the negative exponent as the number of places the decimal point must be moved to the *right* to obtain a number greater than 1 but less than 10: If we begin with 0.0048 and move the decimal point three places to the right, we end up with  $4.8 \times 10^{-3}$ .

In the system of exponential notation, with each shift of the decimal point one place to the right, the exponent *decreases* by 1:

$$4.8 \times 10^{-3} = 48 \times 10^{-4}$$

Similarly, with each shift of the decimal point one place to the left, the exponent *increases* by 1:

$$4.8 \times 10^{-3} = 0.48 \times 10^{-2}$$

Most scientific calculators have a key labeled EXP or EE, which is used to enter numbers in exponential notation. To enter the number  $5.8 \times 10^3$ , the key sequence is

$$\boxed{5} \boxed{\cdot} \boxed{8} \boxed{\text{EXP}} \text{ (or } \boxed{\text{EE}} \text{) } \boxed{3}$$

On some calculators the display will show 5.8, then a space, followed by 03, the exponent. On other calculators, a small 10 is shown with an exponent 3.

To enter a negative exponent, use the key labeled  $+/-$ . For example, to enter the number  $8.6 \times 10^{-5}$ , the key sequence is

$$\boxed{8} \boxed{\cdot} \boxed{6} \boxed{\text{EXP}} \boxed{+/-} \boxed{5}$$

When entering a number in exponential notation, do not key in the 10.

In working with exponents, it is important to recall that  $10^0 = 1$ . The following rules are useful for carrying exponents through calculations.

1. **Addition and Subtraction** In order to add or subtract numbers expressed in exponential notation, the powers of 10 must be the same:

$$\begin{aligned} (5.22 \times 10^4) + (3.21 \times 10^2) &= (522 \times 10^2) + (3.21 \times 10^2) \\ &= 525 \times 10^2 \quad (3 \text{ significant figures}) \\ &= 5.25 \times 10^4 \end{aligned}$$

$$\begin{aligned} (6.25 \times 10^{-2}) - (5.77 \times 10^{-3}) &= (6.25 \times 10^{-2}) - (0.577 \times 10^{-2}) \\ &= 5.67 \times 10^{-2} \quad (3 \text{ significant figures}) \end{aligned}$$

When you use a calculator to add or subtract, you need not be concerned with having numbers with the same exponents because the calculator automatically takes care of this matter.

2. **Multiplication and Division** When numbers expressed in exponential notation are multiplied, the exponents are added; when numbers expressed in exponential notation are divided, the exponent of the denominator is subtracted from the exponent of the numerator:

$$\begin{aligned} (5.4 \times 10^2)(2.1 \times 10^3) &= (5.4)(2.1) \times 10^{2+3} \\ &= 11 \times 10^5 \\ &= 1.1 \times 10^6 \end{aligned}$$

$$(1.2 \times 10^5)(3.22 \times 10^{-3}) = (1.2)(3.22) \times 10^{5-3} = 3.9 \times 10^2$$

$$\frac{3.2 \times 10^5}{6.5 \times 10^2} = \frac{3.2}{6.5} \times 10^{5-2} = 0.49 \times 10^3 = 4.9 \times 10^2$$

$$\frac{5.7 \times 10^7}{8.5 \times 10^{-2}} = \frac{5.7}{8.5} \times 10^{7-(-2)} = 0.67 \times 6.7 \times 10^9 = 6.7 \times 10^8$$

3. **Powers and Roots** When numbers expressed in exponential notation are raised to a power, the exponents are multiplied by the power; when the roots of numbers expressed in exponential notation are taken, the exponents are divided by the root:

$$\begin{aligned} (1.2 \times 10^5)^3 &= (1.2)^3 \times 10^{5 \times 3} \\ &= 1.7 \times 10^{15} \end{aligned}$$

$$\begin{aligned} \sqrt[3]{2.5 \times 10^6} &= \sqrt[3]{2.5} \times 10^{6/3} \\ &= 1.3 \times 10^2 \end{aligned}$$

Scientific calculators usually have keys labeled  $x^2$  and  $\sqrt{x}$  for squaring and taking the square root of a number, respectively. To take higher powers or roots, many calculators have  $y^x$  and  $\sqrt[y]{y}$  (or INV  $y^x$ ) keys.

For example, to perform the operation  $\sqrt[3]{7.5 \times 10^{-4}}$  on such a calculator, you would key in  $7.5 \times 10^{-4}$ , press the  $\sqrt[y]{x}$  key (or the INV and then the  $y^x$  keys), enter the root, 3, and finally press = . The result is  $9.1 \times 10^{-2}$ .

### Sample Exercise 1

Perform each of the following operations, using your calculator where possible:

(a) Write the number 0.0054 in standard exponential notation;

(b)  $(5.0 \times 10^{-2}) + (4.7 \times 10^{-3})$ ;

(c)  $(5.98 \times 10^{12})(2.77 \times 10^{-5})$ ;

(d)  $\sqrt[4]{1.75 \times 10^{-12}}$ .

**SOLUTION** (a) Because we move the decimal three places to the right to convert 0.0054 to 5.4, the exponent is  $-3$ :

$$5.4 \times 10^{-3}$$

Scientific calculators are generally able to convert numbers to exponential notation using one or two keystrokes. Consult your instruction manual to see how this operation is accomplished on your calculator.

(b) To add these numbers longhand, we must convert them to the same exponent:

$$(5.0 \times 10^{-2}) + (0.47 \times 10^{-2}) = (5.0 + 0.47) \times 10^{-2} = 5.5 \times 10^{-2}$$

(Note that the result has only two significant figures.) To perform this operation on a calculator, we enter the first number, strike the + key, then enter the second number and strike the = key.

(c) Performing this operation longhand, we have

$$(5.98 \times 2.77) \times 10^{12-5} = 16.6 \times 10^7 = 1.66 \times 10^8$$

On a scientific calculator, we enter  $5.98 \times 10^{12}$ , press the  $\times$  key, enter  $2.77 \times 10^{-5}$ , and press the = key.

(d) To perform this operation on a calculator, we enter the number, press the  $\sqrt[y]{x}$  key (or the INV and  $y^x$  keys), enter 4, and press the = key. The result is  $1.15 \times 10^{-3}$ .

### Practice Exercise

Perform the following operations: (a) Write 67,000 in exponential notation, showing two significant figures; (b)  $(3.378 \times 10^{-3}) - (4.97 \times 10^{-5})$ ; (c)  $(1.84 \times 10^{15}) / (7.45 \times 10^{-2})$ ; (d)  $(6.67 \times 10^{-8})^3$ . **Answers:** (a)  $6.7 \times 10^4$ ; (b)  $3.328 \times 10^{-3}$ ; (c)  $2.47 \times 10^{16}$ ; (d)  $2.97 \times 10^{-22}$

## A. 2 Logarithms

### Common Logarithms

The common, or base-10, logarithm (abbreviated log) of any number is the power to which 10 must be raised to equal the number. For example, the common logarithm of 1000 (written  $\log 1000$ ) is 3 because raising 10 to the third power gives 1000:

$$10^3 = 1000, \text{ therefore, } \log 1000 = 3$$

Further examples are

$$\log 10^5 = 5$$

$$\log 1 = 0 \quad (\text{Remember that } 10^0 = 1)$$

$$\log 10^{-2} = -2$$

In these examples, the common logarithm can be obtained by inspection. However, it is not possible to obtain the logarithm of a number such as 31.25 by inspection. The logarithm of 31.25 is the number  $x$  that satisfies

$$10^x = 31.25$$

Most electronic calculators have a key labeled LOG that can be used to obtain logarithms. For example, we can obtain the value of  $\log 31.25$  by entering 31.25 and pressing the LOG key. We obtain the following result:

$$\log 31.25 = 1.4949$$

Notice that 31.25 is greater than 10 ( $10^1$ ) and less than 100 ( $10^2$ ). The value for  $\log 31.25$  is accordingly between  $\log 10$  and  $\log 100$ , that is, between 1 and 2.

## Significant Figures and Common Logarithms

For the common logarithm of a measured quantity, the number of digits after the decimal point equals the number of significant figures in the original number. For example, if 23.5 is a measured quantity (three significant figures), then  $\log 23.5 = 1.371$  (three significant figures after the decimal point).

## Antilogarithms

The process of determining the number that corresponds to a certain logarithm is known as obtaining an *antilogarithm*. It is the reverse of taking a logarithm. For example, we saw above that  $\log 23.5 = 1.371$ . This means that the anti-logarithm of 1.371 equals 23.5:

$$\log 23.5 = 1.371$$

$$\text{antilog } 1.371 = 23.5$$

The process of taking the antilog of a number is the same as raising 10 to a power equal to that number:

$$\text{antilog } 1.371 = 10^{1.371} = 23.5$$

Many calculators have a key labeled  $10^x$  that allows you to obtain antilogs directly. On others, it will be necessary to press a key labeled INV (for *inverse*), followed by the LOG key.

## Natural Logarithms

Logarithms based on the number  $e$  are called natural, or base  $e$ , logarithms (abbreviated  $\ln$ ). The natural log of a number is the power to which  $e$  (which has the value 2.71828. . .) must be raised to equal the number. For example, the natural log of 10 equals 2.303:

$$e^{2.303} = 10, \text{ therefore } \ln 10 = 2.303$$

Your calculator probably has a key labeled LN that allows you to obtain natural logarithms. For example, to obtain the natural log of 46.8, you enter 46.8 and press the LN key:

$$\ln 46.8 = 3.846$$

The natural antilog of a number is  $e$  raised to a power equal to that number. If your calculator can calculate natural logs, it will also be able to calculate natural antilogs. On some calculators, there is a key labeled  $e^x$  that allows you to calculate natural antilogs directly; on others, it will be necessary to first press the INV key followed by the LN key. For example, the natural antilog of 1.679 is given by

$$\text{Natural antilog } 1.679 = e^{1.679} = 5.36$$

The relation between common and natural logarithms is as follows:

$$\ln a = 2.303 \log a$$

Notice that the factor relating the two, 2.303, is the natural log of 10, which we calculated above.

## Mathematical Operations Using Logarithms

Because logarithms are exponents, mathematical operations involving logarithms follow the rules for the use of exponents. For example, the product of  $z^a$  and  $z^b$  (where  $z$  is any number) is given by

$$z^a \cdot z^b = z^{(a+b)}$$

Similarly, the logarithm (either common or natural) of a product equals the *sum* of the logs of the individual numbers:

$$\log ab = \log a + \log b \quad \ln ab = \ln a + \ln b$$

For the log of a quotient:

$$\log (a/b) = \log a - \log b \quad \ln (a/b) = \ln a - \ln b$$

Using the properties of exponents, we can also derive the rules for the logarithm of a number raised to a certain power:

$$\begin{aligned} \log a^n &= n \log a & \ln a^n &= n \ln a \\ \log a^{1/n} &= (1/n) \log a & \ln a^{1/n} &= (1/n) \ln a \end{aligned}$$

## pH Problems

One of the most frequent uses for common logarithms in general chemistry is in working pH problems. The pH is defined as  $-\log [\text{H}^+]$ , where  $[\text{H}^+]$  is the hydrogen ion concentration of a solution (Section 16.3). The following sample exercise illustrates this application.

### Sample Exercise 2

- What is the pH of a solution whose hydrogen ion concentration is 0.015 M?
- If the pH of a solution is 3.80, what is its hydrogen ion concentration?

**SOLUTION** (a) We are given the value of  $[H^+]$ . We use the LOG key of our calculator to calculate the value of  $\log [H^+]$ . The pH is obtained by changing the sign of the value obtained. (Be sure to change the sign *after* taking the logarithm.):

$$[H^+] = 0.015$$

$$\log [H^+] = -1.82 \quad (2 \text{ significant figures})$$

$$\text{pH} = -(-1.82) = 1.82$$

(b) To obtain the hydrogen ion concentration when given the pH, we must take the antilog of  $-\text{pH}$ :

$$\text{pH} = -\log [H^+] = 3.80$$

$$\log [H^+] = -3.80$$

$$[H^+] = \text{antilog}(-3.80) = 10^{-3.80} = 1.6 \times 10^{-4} \text{ M}$$

### Practice Exercise

Perform the following operations: (a)  $\log (2.5 \times 10^{-5})$ ; (b)  $\ln 32.7$ ; (c) antilog  $-3.47$ ; (d)  $e^{-1.89}$ . **Answers:** (a)  $-4.60$ ; (b)  $3.487$ ; (c)  $3.4 \times 10^{-4}$ ; (d)  $1.5 \times 10^{-1}$

## A.3 Quadratic Equations

An algebraic equation of the form  $ax^2 + bx + c = 0$  is called a *quadratic equation*. The two solutions to such an equation are given by the quadratic formula:

$$x = \frac{-b \pm \sqrt{b^2 - 4ac}}{2a}$$

### Sample Exercise 3

Find the values of  $x$  that satisfy the equation  $2x^2 + 4x = 1$ .

**SOLUTION** To solve the given equation for  $x$ , we must first put it in the form

$$ax^2 + bx + c = 0$$

and then use the quadratic formula. If

$$2x^2 + 4x = 1$$

then

$$2x^2 + 4x - 1 = 0$$

Using the quadratic formula, where  $a = 2$ ,  $b = 4$ , and  $c = -1$ , we have

$$\begin{aligned} x &= \frac{-4 \pm \sqrt{(4)(4) - 4(2)(-1)}}{2(2)} \\ &= \frac{-4 \pm \sqrt{16 + 8}}{4} = \frac{-4 \pm \sqrt{24}}{4} = \frac{-4 \pm 4.899}{4} \end{aligned}$$

The two solutions are

$$x = \frac{0.899}{4} = 0.225 \quad \text{and} \quad x = \frac{-8.899}{4} = -2.225$$

Often in chemical problems the negative solution has no physical meaning, and only the positive answer is used.

**TABLE 1 Interrelation  
Between Pressure and  
Temperature**

Temperature (°C)	Pressure (atm)
20.0	0.120
30.0	0.124
40.0	0.128
50.0	0.132

## A.4 Graphs

Often the clearest way to represent the interrelationship between two variables is to graph them. Usually, the variable that is being experimentally varied, called the *independent variable*, is shown along the horizontal axis ( $x$  axis). The variable that responds to the change in the independent variable, called the *dependent variable*, is then shown along the vertical axis ( $y$  axis). For example, consider an experiment in which we vary the temperature of an enclosed gas and measure its pressure. The independent variable is temperature, and the dependent variable is pressure. The data shown in Table 1 can be obtained by means of this experiment. These data are shown graphically in Figure 1. The relationship between temperature and pressure is linear. The equation for any straight-line graph has the form

$$y = mx + b$$

where  $m$  is the slope of the line, and  $b$  is the intercept with the  $y$  axis. In the case of Figure 1, we could say that the relationship between temperature and pressure takes the form

$$P = mT + b$$

where  $P$  is pressure in atm and  $T$  is temperature in °C. As shown in Figure 1, the slope is  $4.10 \times 10^{-4}$  atm/°C, and the intercept—the point where the line crosses the  $y$  axis—is 0.112 atm. Therefore, the equation for the line is

$$P = \left(4.10 \times 10^{-4} \frac{\text{atm}}{^\circ\text{C}}\right) T + 0.112 \text{ atm}$$

► FIGURE 1

